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Unravelling the room temperature growth of two-dimensional h-BN nanosheets for multifunctional applications[†]

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The room temperature growth of two-dimensional van der Waals (2D-vdW) materials is indispensable for state-of-the-art nanotechnology. Low temperature growth supersedes the requirement of elevated growth temperatures accompanied with high thermal budgets. Moreover, for electronic applications, low or room temperature growth reduces the possibility of intrinsic film-substrate interfacial thermal diffusion related deterioration of the functional properties and the consequent deterioration of the device performance. Here, we demonstrated the growth of ultrawide-bandgap boron nitride (BN) at room temperature by using the pulsed laser deposition (PLD) process, which exhibited various functional properties for potential applications. Comprehensive chemical, spectroscopic and microscopic characterizations confirmed the growth of ordered nanosheet-like hexagonal BN (h-BN). Functionally, the nanosheets show hydrophobicity, high lubricity (low coefficient of friction), and a low refractive index within the visible to near-infrared wavelength range, and room temperature single-photon quantum emission. Our work unveils an important step that brings a plethora of potential applications for these room temperature grown h-BN nanosheets as the synthesis can be feasible on any given substrate, thus creating a scenario for "h-BN on demand" under a frugal thermal budget.

Introduction

Two-dimensional van der Waals (2D-vdW) materials are astonishing in the nano-era, showing tremendous potential and

New concepts

The growth of two-dimensional van der Waals materials at low/room temperature is indispensable for state-of-the-art nanotechnology as it supersedes the requirement of an elevated growth temperature that comes with additional high thermal budgets (and with the possibility of an increase of interfacial thermal diffusion). Here, two-dimensional hexagonal boron nitride (h-BN) nanosheets are grown at room temperature by a highly energetic pulsed laser deposition (PLD) growth technique, exhibiting remarkable functional properties and suitability for a number of applications (e.g. hydrophobicity, a low refractive index in the visible wavelength range, low frictional force, and remarkable room temperature single-photon emission), thus creating a potential scenario for "h-BN on demand" under a frugal thermal budget, which is indispensable for nanotechnology.

technological relevance due to their unique atomic-scale growth, emergent functional properties, and application potential.¹⁻⁴ In general, several top-down (chemical, and mechanical exfoliations) and bottom-up approaches (thin films growth by various physical/chemical vapor phase deposition techniques) have been employed to synthesize these 2D-vdW materials with atomic layer control.⁵⁻⁷ Among these, the liquid exfoliation methods produce large-area 2D-vdW materials; however, the film quality is found to be very low (due to the presence of adsorbates and surfactants on the surface). Whereas vapor-phase thin film growth processes lead to smaller sized crystals, but with high crystalline quality.⁸ Due to this process-dependent variable quality, the growth of 2D-vdW materials thus remains an

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exceedingly intriguing research topic among the materials growth community. The formation, crystallinity, morphology, and consequent functional properties of 2D-vdW materials are highly sensitive to the thermodynamic and/or growth kinetics, the adsorption of the reactive species on the substrate surface, nucleation, and thus the resultant film growth and its consequent properties.^{9,10}

Among various procedures, for large-scale growth, intrinsic property evaluation, and nano-device fabrication, chemical vapor deposition (CVD) has widely been adopted to grow 2DvdW materials.^{5,6} However, synthesizing 2D-vdW materials by CVD requires a very high-temperature of \sim 1000–1500 °C (for a sufficient supply of activation energy to the adatoms so that they can migrate to the energetically preferred locations on the substrates during growth); a rigorous synthesizing condition that limits device capabilities as elevated temperatures result in possible defect segregation and annihilation at the interfaces.¹¹⁻¹⁴ Therefore, researchers are trying to synthesize 2D-vdW materials at relatively low-temperatures, with a trade-off between crystalline domain size/quality and generating novel application possibilities.¹⁵⁻¹⁸ Theoretically, 2D materials show anisotropy in growth and kinetics that can be controlled by adjusting the balance between elements.¹⁹ However, fundamentally, both thermodynamics (temperature) and kinetics play important roles, therefore the temperature during the vapor-phase deposition has always been critical, which also affects the nucleation density and grain size.^{10,15} Therefore, the growth of 2D-vdW materials at lower temperatures or even at room temperature with a high-enough supply of kinetic energy directed towards the reaction species might be sufficient to overcome the nucleation barrier, and if successful, it would be very useful for several room-temperature applications. This would significantly enable the growth of 2D-vdW materials on any given substrate (with varying degrees of crystallinity, thickness, and uniformity and their consequent emergent phenomena), creating a scenario for "thin films growth of 2D-vdW materials on demand", thereby leading to ample investigations of their properties and their consequent applications.

To investigate this possibility of thin film room-temperature growth, we use hexagonal boron nitride (h-BN) as our model material. Among numerous 2D-vdW materials, ultra-wide bandgap h-BN (with a bandgap of \sim 5.9 eV) gets special attention for its excellent chemical inertness, high-thermal stability, and electronic, optical, and mechanical properties.^{20,21} Electronically, h-BN is suitable as a gate-dielectric layer for 2D-based field effect transistor (FET) devices.²² The high bandgap of h-BN makes it suitable for the fabrication of high-performance deep-ultraviolet photodetectors.²³ Moreover, the weak inter-layer bonding in h-BN makes it a soft lubricant material and thus useful as hightemperature corrosion resistant and antioxidation protective coatings for various industrial applications.²¹ Given these advantages, we attempt to grow h-BN thin films at room temperature by using a highly energetic and thermally non-equilibrium thin film growth process, pulsed laser deposition (PLD). PLD possesses several advantages over other growth methods as it is not in thermal equilibrium, directed onto the substrate, and the

stoichiometry is preserved throughout the thin film deposition process, starting from a dense crystalline target of the desired material.²⁴ Several attempts have been implemented to grow h-BN films by PLD; however, at elevated temperatures, they show mostly island-like growth.^{25–30} The plasma formation during the ablation process of the target by pulsed UV-laser (photon energy of ~5 eV) contains radicals and ions (*e.g.* B⁺, N⁺, N*, N₂, N₂⁺, N₂* and B*N⁺) of the ablated species with kinetic energy ~10–100 eV (Fig. 1a and b).³¹

We have grown h-BN at room temperature on various substrates: c-Al₂O₃ (0001), a holey Cu-grid, and Si (100), and investigated their growth characteristics and several functional properties. Chemical, morphological, spectroscopic, and electron microscopy characterizations confirm the growth of ordered h-BN nano-sheets. We observed that the films are hydrophobic, and exhibit a low-refractive index, and room temperature singlephoton quantum emission. Moreover, the frictional characteristics reveal excellent lubrication properties with a low coefficient of friction. Thus, successful growth of ordered h-BN nano-sheets by PLD at room temperature might lead to the growth of h-BN on a variety of substrates, generating ample potential applications under reduced thermal budgets, e.g. for flexible 2D-electronics,³² coating layers for highly-corrosive materials, protecting layers for open-air degradable materials, precise photonic devices and for quantum information technology.

Results and discussion

Structurally, to characterize the grown h-BN films we first performed conventional B 1s and N 1s core-level X-ray photoelectron spectroscopy (XPS) elemental scans. Characteristically, we observed the B-N bonding-related peaks in both the elemental B 1s (at \sim 190.6 eV) and N 1s (at \sim 398.5 eV) core level scans with π -plasmon peaks (~9 eV apart from the main B-N peak), which are characteristic of h-BN (Fig. 1c, d and Fig. S1a, S1b, ESI⁺).³³ In the N 1s scans, in addition to the B-N bonding peak, we observed a small shoulder of the N-C peak at \sim 400.5 eV (Fig. 1d), due to the ambient air exposure effect, related to adventitious carbon.³⁴ Through the valence band spectra (VBS), we observed the characteristic band structure of the h-BN film (two distinct features with maxima at \sim 12 and \sim 20 eV, corresponding to the overlapping of the σ and π -bands as well as the *s*band, respectively) with a valence band maxima (VBM) position at ~1.7 eV (Fig. S1c, ESI^{\dagger}).^{35,36} In order to confirm the phase, we then performed Fourier transform infrared (FTIR) spectroscopy and Raman spectroscopy. FTIR shows a peak within the range 1350-1500 cm⁻¹, attributed to the in-plane B-N stretching vibrations (transverse optical mode) of the sp²-bonded h-BN (Fig. 1e).³⁷ Moreover, in the Raman spectra we observed a Raman-active in-plane E_{2g} phonon mode at ~1364 cm⁻¹ (Fig. 1f),³³ with a narrow full-width at half maximum (FWHM) of ~14.85 cm⁻¹, along with peaks at \sim 577.4 and \sim 749.2 cm⁻¹, originating from the sapphire substrate. The Raman peak is slightly red shifted $(\sim 2 \text{ eV})$ to lower phonon frequencies compared to the bulk h-BN (\sim 1366 eV), suggesting that the nanosheets are showing slight



Fig. 1 Pulsed laser deposition of h-BN and its spectroscopic characterizations. (a) Schematic showing the thin film growth process by pulsed laser deposition. (b) Real-time plume generation (highly energetic) during the ablation of h-BN with ablated species deposited on a substrate kept at room temperature. An optical image of a uniform ($\sim 1 \times 1 \text{ cm}^2$) thick h-BN film (bottom). (c) and (d) XPS elemental B 1s and N 1s core scans on c-Al₂O₃ confirm the presence of characteristics B–N bonds. (e), (f) FTIR and Raman spectra shows the peak at $\sim 1364 \text{ cm}^{-1}$, further confirming the growth of h-BN at room temperature.

tensile strain.³³ All these spectroscopic characterizations indicate the room-temperature growth of h-BN films.

Hereafter, we investigated the surface morphology of the films by using atomic force microscopy (AFM). Interestingly, they show 2D nanosheet-like morphology (on *c*-Al₂O₃) (Fig. 2a). The lateral sizes of these h-BN sheets are ~200–300 nm and the area coverage per $3 \times 3 \mu m^2$ is nearly ~70 441.75 nm² (Fig. S2, ESI†). Furthermore, to confirm the crystalline nature, we performed top-view high-resolution transmission electron microscopy (HRTEM) by growing h-BN films directly on a holey Cu grid.

For HRTEM, we have grown ultra-thin h-BN by providing only a small number of laser shots (100 laser shots to be precise). As seen from the HRTEM images (Fig. 2b), we also observed nanocrystal-like features. We captured several images at different magnification levels and obtained the diffraction patterns and *d* spacing. Remarkably, a closer inspection shows the appearance of clear ordered lattice fringes (Fig. 2c). We obtained the *d* spacing which corresponds to the vdW gap between individual h-BN sheets ($d_{0002} = 0.33$ nm) with a hexagonal diffraction pattern, although the signals are very weak in intensity



Fig. 2 Microscopic characterizations of h-BN. (a) AFM topography shows the nanosheet-like growth. (b) and (c) High-resolution transmission electron microscopy (HRTEM) also shows the sheet-like structure with the corresponding diffraction pattern (inset). (d) Inter-planar *d* spacing of \sim 0.33 nm, corresponding to the (0002) lattice plane of h-BN. For HRTEM, a few shots of BN were supplied directly on the holey Cu grid.

(Fig. 2d and Fig. S3, ESI⁺).^{20,21} The TEM images, corresponding diffraction patterns and *d*-spacing also show that along with the ordered hexagonal structure, a degree of amorphization (evident from the amorphous disk in the diffraction pattern) also exists in the nanosheets, at least for the ultrathin h-BN grown on the holey Cu-grid at room temperature.

As mentioned earlier, the room temperature growth of highly ordered nanosheets is extremely interesting and costeffective for scale-up, as they can be grown on potentially any given substrate (e.g. foils, flexible polymer surfaces), widening the horizon of h-BN growth for applications.²⁰ There have been attempts to grow h-BN at room temperature by RF sputtering and atomic layer deposition (ALD), but these attempts have formed an amorphous/turbostratic phase.^{38,39} In the case of PLD, which is most often used for metal-oxide thin film depositions, it is well known that a high-temperature (\geq 700 °C) is needed for epitaxial growth (depending on the thermodynamics).²⁴ However, as reported by Elhamid et al., well-defined graphene can be grown at room temperature by PLD on a bi-metallic Ni-Cu substrate, where the laser power played a crucial role, favoring a nucleation activation energy which promotes graphene formation.⁴⁰ Rasic et al. grew metallic TiN thin films on sapphire at roomtemperature.41 Kakehi et al. reported the epitaxial growth of NiO (111) thin films by PLD at room-temperature.⁴² Ma et al. grew ZnO

films on glass at room temperature by PLD.⁴³ Fundamentally, the adsorption energy is one order higher than the diffusion energy barrier and at lower temperature, the adsorption rate exceeds the desorption rate.¹⁵ Therefore, small-fractal shaped domain growth has been observed for the vapor-phase synthesis of 2D materials at high flux and reduced temperature.¹⁵ Very recently, people have also started putting in effort to grow h-BN at various temperatures by PLD.⁴⁴ Hence, from a growth perspective, an energetic bombardment of particles during the growth process is necessary to synthesize h-BN films at lower/room temperature, with the ability for the adatoms to overcome the substrate diffusion energy barrier. Ideally, the thin film synthesis process involves: (1) evaporation and transport of the ablated species to the substrate, (2) migration of the ablated species onto the substrate, and (3) nucleation and crystal growth.²⁴ For h-BN growth, the laser ablated species are atomic neutral and ionized species including B⁺, B^{*}, N⁺, and N^{*}, and molecular species including N₂^{*}, N₂⁺, and B*N⁺.³¹ These species arrive concurrently on the substrate with a high kinetic energy of \sim 10–100 eV, and thereby form h-BN layers (nanosheets in our case) within microseconds ($\sim 2 \mu S$) onto the substrate, even at room temperature. The importance of the substrate temperature is to supply additional energy for these species to mobilize onto the surface, coalesce, form bonds with the substrates, thus promoting better nucleation, and consequent



Fig. 3 Functional properties of h-BN. (a) Contact angle measurements on the h-BN surface, showing the water repelling nature. (b) Topography (upper panel) and the corresponding frictional force (lower panel) map of h-BN/c-Al₂O₃. The darker colors in the frictional force map of the h-BN suggest lower friction. (c) The measured frictional force with different loading. The slope of the fitted line represents the coefficient of friction (COF) of the material. (d) The refractive index (*n*) of the h-BN film (grown on Si substrate). The inset shows the schematic of the measurement.

crystallinity. We have also performed FESEM and grazing incident angle X-ray diffraction (GIXRD) of the h-BN target, before and after the laser ablation (Fig. S4, ESI[†]). In both cases, the FESEM images show similar morphology with a typical sheet size of several µm, which is much larger than the lateral sizes for the room temperature grown h-BN nanosheets, as shown in the AFM images (Fig. 3b). The GIXRD results show only the reduction in the (0002) h-BN peak intensity after laser ablation, which is apparent as the target surface becomes rougher after the ablation. Considering PLD is a highly energetic deposition process and in addition BN contains light elements, thus synergistically, created vapor pressure and kinetic energy of the laser-ablated species in the plasma form possibly sufficient to migrate to the substrate and nucleate onto it (within a few μ S) and forms the films even at room temperature, affecting the nucleation density and grain size.15,24 All these observations reduce the possibility that growth happens due to the sputtering of nano-BN particles from the target, but rather it follows the typical thin film growth process, at least for the present employed growth protocols.

To realize the intrinsic functional properties of the h-BN nanosheets, first, we investigate the wetting characteristics by contact angle (CA) measurements, as hydrophobic surfaces are known as effective protective layers in harsh environments with extreme temperatures, and h-BN provides strong resistance to chemical attack and high-temperature stability.²⁰ As observed, h-BN tends toward minimizing the contact surface with water.

The nanosheets exhibit hydrophobic behavior which is fairly increased by ~5%, (CA: $55^{\circ} \pm 1$ to $60^{\circ} \pm 1$) (Fig. 3a), consistent with the polar character of the boron-nitrogen bond ($-N^-B^+-$), as the wettability is strongly influenced by the vdW interactions.⁴⁵ The measured CA is also in agreement with the reported values ranging between ~50–67°.⁴⁵ Fundamentally, the wetting characteristics of a particular material also depend on the surface roughness and are inherent to the wetting phenomena.^{46,47} However, considering that the surface roughness of the grown h-BN nanosheets is of a few nm (~2–3 nm), the effect of roughness on the CA may be insignificant here.

The effects of the hydrophobicity and inertness of the h-BN nanosheets are further investigated through force-distance (F-D) spectroscopy measurements and the associated line profile using an AFM probe covered with diamond-like carbon (DLC) (Fig. S5 and S6, ESI[†]). In the F-D measurements, the AFM probe temporarily brings a closer "repulsive regime" to the h-BN nanosheets for short durations (sub-seconds) and then moves away vertically from the surface. It requires work against the adhesion force that brings the cantilever away from the surface (*i.e.* h-BN nanosheet and *c*-Al₂O₃), referred to as a "pull-out" force. A higher value of the pull-out force indicates a vital interaction with the substrate towards the tip apex and *vice versa*. The h-BN nanosheets show smaller pull-out force values towards the DLC tip as compared to the host substrate corroborating the CA measurements.

At the submicron scale, the intrinsic roughness and moderately hydrophobic surface influence the tribological response. Thus, the frictional characteristics of h-BN (on c-Al₂O₃) are examined by lateral force microscopy where the topography and the corresponding frictional force map are measured at the same acquisition (Fig. 3b). The presence of h-BN layers may lead to the prediction of lower friction force values than the surrounded c-Al₂O₃ substrate. However, the edges and stepedges lead to the highest friction values due to the higher susceptibility of edge atoms towards the slider (tip apex) which is peculiar to the family of 2D-vdW materials.^{48,49} The tendency of the lubrication is further monitored where the probe sweeps the same region of h-BN and c-Al₂O₃ with an increasing applied normal force ranging from 10 nN to 350 nN (Fig. 3c). We observed a linear response of the increase in the friction force as a function of the normal forces, where the trend of loaddependent friction is higher for the alumina substrate as compared to h-BN, which is quantified through a linear fit resulting in the coefficient of friction (COF). The COF values of h-BN and c-Al₂O₃ are obtained as ~0.0002 and ~0.0011, respectively. The COF value depends on the nature of the material used and the local contact area between the tip apex and the substrate. Considering that the local contact area of the tip apex is similar while rubbing between h-BN and alumina surface in a single scan at fixed load conditions, it is the nature of the material dominating the higher COF values for alumina. The additional factor influencing the frictional behavior in 2D materials is the interfacial interaction with the host surface. It is observed that the c-Al₂O₃ plane is relatively reactive w.r.t. the other planes (e.g. a-Al₂O₃), to establish a stronger interfacial bond with h-BN.⁵⁰ Weak vdW interactions (e.g. h-BN-silica, graphene-silica) lead to higher friction forces of the atomically thin sheets owing to the puckering effect "i.e. out-off plane deformation that resists sliding" compared to the thick layers.49 Whereas, a stronger interfacial interaction (*i.e.* graphene-mica, and graphene-Ni(111)) does not show the puckering effect, revealing that the frictional force response is independent of the thickness.^{51,52} We did not observe a thickness dependent frictional contrast in our friction map, indicating a stronger interaction between h-BN and c-Al₂O₃ (Fig. S2, ESI[†]). These results are in good agreement with conventionally fabricated h-BN nanosheets and other 2D materials.52-54 Overall, the observed hydrophobic nature and lubricity of the h-BN nanosheets would be useful for the coating industry.

Furthermore, we measured the refractive index (*n*) of a h-BN nanosheets grown on a Si substrate (see ESI[†] Fig. S7 for the detailed characterizations of the film), which was found to be $\sim 1.53-1.55$ (Fig. 3d) within the visible and near-infrared spectral range. We used variable angle spectroscopic ellipsometry (VASE) to measure the refractive index (*n*). The schematic of the measurement is depicted in the inset of Fig. 3d. Considering the nanosheet-like morphology of h-BN, we fitted the ellipsometry data by applying the Bruggeman effective-media approximation (EMA) method that considers the film to be h-BN with defects (see the methods section and Fig. S8, ESI[†]). Interestingly, the h-BN with added voids into the fitting model provides a

lower *n* than the perfect single crystalline layer h-BN fitting $(n \sim 1.8)$. The lower *n* is adequate as stoichiometry, porosity, and roughness affect the value of n.^{55,56} Interestingly, the lower *n* is helpful for the precise designing of photonic devices operating in the visible and near-infra-red wavelength range.⁵⁵

Since h-BN nanosheets can host defects such as point defects (including boron-vacancies (V_B) and/or nitrogen-vacancies (V_N)), nitrogen-vacancies and boron replacing nitrogen (V_NN_B), and also carbon impurities, they may show single photon emission (SPE) at room temperature, attributed to these localized defect states inside the bandgap.^{57–61} SPE in h-BN is extremely important for next-generation quantum computing and information-processing technologies, considering its excellent chemical and thermal robustness.^{62,63} Fig. 4a shows a confocal photoluminescence (PL) map (~50 × 50 µm²) of h-BN nanosheets showing much brighter emission spots (yellow) above the background.

Fig. 4b shows the corresponding PL emission spectrum under continuous wave non-resonant excitation ($\lambda = 532$ nm). The emitter has a sharp zero-phonon line (ZPL) at ~577 nm with an asymmetric shape (FWHM ~2 nm), extracted from a single Lorentzian fit. A weak optical phonon replica is visible around ~630 nm, and is red-shifted by ≥ 165 nm from the ZPL peak, which is related to the phonon sidebands (PSB).⁶⁴ It should be noted that the small peak at ~580 nm is most likely a ZPL of another weak emitter within the excitation spot. In order to understand the extent of the electron–phonon coupling we calculate the Debye–Waller factor (the ratio between ZPL intensity to the total emission) which is found to be ~0.70.

In order to understand the polarization behavior of the localized emission, we performed emission polarization measurements. Fig. 4c corresponds to the angular diagram of emission, which reflects a well-defined dipole-like character for the single isolated defect. The corresponding fits are obtained using a $\cos^2(\theta)$ fitting function, which yields an emission polarization visibility of \sim 75%, suggesting a single dipole transition linearly oriented along the basal plane of the h-BN nanocrystals. Fig. 4d displays the second-order correlation function $g^2(\tau)$ using two Si APDs, mounted in a Hanbury-Brown and Twiss (HBT) configuration to establish the quantum nature of the emitted light from the defects of the h-BN nanocrystals. The $g^2(0)$ is found to be ~0.36, well below 0.5 (without background correction) at zero delay time (Fig. 4d), which clearly indicates the anti-bunching of the emission.⁶² The experimental $g^2(\tau)$ data can be fitted well with the threelevel model: $g^{2}(\tau) = 1 - (1 + a)e^{-|\tau|/\tau} + ae^{-|\tau|/\tau}$, where *a* is a fitting parameter, and τ_1 and τ_2 are the life-times of the excited and metastable states, respectively. Moreover, the radiative lifetime $\tau_1 \sim 3$ ns is consistent with that of the previous report,⁶⁴ however the non-radiative lifetime τ_2 is much longer. These observations clearly show the room temperature stable SPE in h-BN. Our results indicate that this stable and bright SPE observed in the visible range at room temperature could potentially solve the long-term issues of the on-chip integration of deterministic SPEs over large scales. Our approach of growing h-BN nanosheets even at room temperature depicts the ability to



Fig. 4 Room-temperature photo-physical properties of h-BN nanosheets. (a) Confocal photoluminescence map ($\sim 50 \times 50 \ \mu m^2$) corresponds to the ultra-bright (yellow spots) defect-assisted emission from the h-BN nano-sheets grown on sapphire. (b) Photoluminescence spectrum under 532 nm laser excitation and a 550 nm long-pass filter at room temperature. (c) The emission (red spheres) polarization curve. The red curve is fitted using a $\cos^2\theta$ function indicating an atomic defect with the in-plane dipole moment. (d) Second-order autocorrelation measurement $g^{(2)}(\tau)$ of the defect with a dip at $g^{(2)}(0) \approx 0.36$, suggesting the antibunching nature of light.

directly grow 2D layers onto designed electronic/photonic structures, which can be very useful for integrated circuit applications since it avoids any chemical contaminants or material deformation that often occur with wet or dry transfer.

In view of device applications, h-BN nanosheets show excellent promise. 2D h-BN has emerged as a suitable material for the fabrication of cost-effective deep-ultraviolet (UV) photodetectors.⁶⁵ Recently, it was shown that h-BN nanosheets could be useful as a saturable absorber for mode-locked lasers.⁶⁶ h-BN can act as a surface passivation layer for reducing the electrical loss of 2D-vdW based solar cells.⁶⁷ h-BN could also be useful as a tunnel barrier in 2D-vdW heterostructures and devices for information science and nanotechnology.⁶⁸ Considering the fact that h-BN could be grown on Si at room temperature, this would solve the thermal stability related issues of Si substrates, and the high-temperature growth of h-BN on Si would be advantageous for the design of Si-based integrated photonics.⁶⁹ For energy harvesting, h-BN nanosheet based devices are also useful to develop a high-performance hybrid piezo/triboelectric nanogenerator (PTEG) by incorporating it into polydimethylsiloxane (PDMS).⁷⁰ Therefore, room temperature in situ grown chemical contaminant free h-BN nanosheets would be extremely beneficial for next-generation cost-effective clean optoelectronic and energy harvesting applications.

Conclusion

In summary, we have shown that by using a highly energetic pulsed laser deposition (PLD) vapor phase thin film growth process, ordered nanosheets of 2D-vdW h-BN could be grown on desirable substrates, and remarkably, even at room temperature. This is attributed to the ability of the adatoms to overcome the substrate diffusion energy barrier, because of the interplay between the thermodynamics and the high kinetics of the PLD process. We used extensive characterizations including chemical, spectroscopic, microscopic, tribological and optical tools to confirm the growth of the ordered 2D-vdW h-BN nanosheets with the exhibition of various properties. Functionally, the h-BN nanosheets show improved-hydrophobicity and a low refractive index in the visible wavelength range. Furthermore, they show excellent lubricity with a reduction in the friction force coefficient compared to that of the subsurface. Astonishingly, the films also exhibit room temperature single-photon emission, an important building block of optical quantum nanotechnologies. Therefore, considering their suitability for a number of applications, our observation of the successful growth of 2D-vdW h-BN nanosheets at room temperature, might lead to their growth on any substrate (and might also lead the way to the growth of other 2D-vdW materials at room temperature), generating ample potential applications under reduced thermal budgets, e.g. flexible

2D-electronics, coating layers for highly corrosive and degradable materials, precise photonic devices and for quantum information technology, thus potentially creating a scenario of "*h-BN on demand*" under a frugal thermal budget, which is indispensable for nanotechnology applications.

Methods

Thin film growth (pulsed laser deposition)

The h-BN nanosheets were grown by PLD (KrF excimer laser with an operating wavelength of 248 nm, and a pulse width of 25 ns). The films were grown at room-temperature by using the following deposition conditions: laser fluency ~2.2 J cm⁻² (laser energy ~230 mJ, spot size ~ 7 mm × 1.5 mm), repetition rate of 5 Hz, high-purity (5N) nitrogen gas with a partial pressure (P_{N2}) ~ 100 mTorr (flow rate ~73 sccm), and a target to substrate distance ~50 mm. We used a commercially available one-inch dense high-purity (99.9% metal basis) polycrystalline h-BN target for the ablation (American Element). The chamber base pressure was 5 × 10⁻⁹ Torr. For the substrates, we used hexagonal *c*-Al₂O₃ (0001), Si (100), and a holey Cu grid, and 2000 laser pulses (on Al₂O₃), 5000 laser pulses (on Si) and 100 laser pulses (on the holey Cu grid) were given to the target, respectively. The substrates were purchased from MSE suppliers, USA.

Chemical and microscopic characterizations (XPS, VBS, FTIR, AFM, Raman, GIXRD, FESEM, and HRTEM)

X-ray photoelectron spectroscopy was performed using a PHI Quantera SXM scanning X-ray microprobe with a 1486.6 eV monochromatic Al K_{α} X-ray source. High-resolution scans were recorded at 26 eV pass energy. The valence band spectra were recorded at 69 eV pass energy. XPS curve fitting was also performed using a Shirley baseline with the Multipak XPS software. FTIR was obtained by using a Nicolet 380 FTIR spectrometer (4000-400 cm⁻¹ region), equipped with a singlecrystal diamond window (resolution 8 cm⁻¹). A Park NX20 AFM was used to obtain the surface topography, operating in tapping mode using Al-coated Multi75Al cantilevers. Raman spectroscopy was performed using a Renishaw inVia confocal microscope, with a 532 nm (50 \times resolution, 10% power) laser as the excitation source. The grazing incident angle X-ray diffraction (GIXRD) (with incident angle of 0.1°) was obtained using a Rigaku SmartLab X-ray diffractometer, with a monochromatic Cu K α radiation source (λ = 1.5406 Å). The h-BN target surface topography was performed by field emission scanning electron microscopy (FESEM) (FEI Quanta 400 ESEM FEG). We sputtered \sim 10 nm gold on h-BN for the FESEM measurements. For the HRTEM measurements, the ultra-thin film was directly grown on a holey Cu-grid and the images were recorded using a Titan Themis instrument operating at 300 kV and the images were obtained at different magnification levels.

Adhesion and friction force characterizations

Atomic force microscopy (from Bruker Ltd) was conducted using a Bruker ScanAsyst instrument using the peak force tapping mode (Bruker's proprietary) with an applied normal force of 1 nN. All the AFM operations were carried out at room temperature under \sim 35–40% of the relative humidity. Several AFM tip materials (i.e. cantilever) were used in the measurements to investigate the surface chemistry (adhesion force) between the h-BN, substrate and tip apex. A silicon nitride cantilever (model: Scanasyst-air (Bruker), stiffness \sim 0.5 \pm 0.15 N m⁻¹) was used for the topography measurements. A diamond-like carbon coated tip (model: DT-CONTR (Nanosensors), stiffness $\sim 40 \pm 2$ N m⁻¹) was used for the forcedistance spectroscopy and mechanical mapping. The friction force (nN) was measured in the mode of "lateral force microscopy", and a silicon cantilever (model: CSC38 (Micromash) of stiffness (0.14 \pm 0.05) N m⁻¹ by Sader's method) was used for the investigations.⁷¹ The average values of the lateral forces in the trace and retrace directions are measured as half of the trace minus retrace (TMR) values. This procedure is useful to mitigate the topographical effects in the friction measurements. The tip radii are evaluated through the blind tip reconstruction technique using a grating sample as described in previous studies.72

Optical characterizations (refractive index and single-photon emission)

Variable angle spectroscopic ellipsometry (VASE) (M-2000 Ellipsometer by J. A. Woollam Company) was applied to measure the refractive index (RI). The RI of a bare silicon substrate with a thin native oxide layer on the top is measured first as a reference, followed by the RI of the film. Two sequential measurements followed a similar process. The incidents of four different angles equally spaced from $\Phi = 55^{\circ}$ to 70° were sequentially shed on the surface of the samples that were placed on a horizontal stage and the corresponding reflected light was collected by a detector. The original spectroscopic data were then fitted using two types of three-layer models (see the Fig. S8 for their detailed comparison, ESI†).

In both models, a two-layer reference model that consists of a pure silicon substrate beneath a 1.72 nm native oxide layer was applied to represent the silicon substrate. Then a third layer was added on top of the reference model to represent the grown h-BN film. We used a Bruggeman effective-media approximation (EMA) layer that mixes the pure h-BN with voids (n = 1). The light spot generated by the ellipsometer has a diameter of a scale of hundreds of microns, which is large enough to bring an average effect for the h-BN film. We also used the pure h-BN part as the third layer. The fitting result of this model gave an unacceptable MSE greater than 600, which was not a suitable fitting parameter.

We performed single photon emission at room temperature by using a conventional confocal, optical microscope and a Hanbury-Brown and Twiss (HBT) interferometer. The emitted light is collected by an air objective ($100 \times$, NA 0.95) and detected by using avalanche photodiodes or a spectrometer. A 532 nm CW laser diode was used to excite the SPEs. A dichroic mirror (cutoff wavelength 532 nm) and a long pass filter (cutoff wavelength 550 nm) allowed the suppression of the back-reflected **Nanoscale Horizons**

pump beam. The signal was fiber-coupled to either a grating spectrometer (Ocean optics, QE65pro) or avalanche photodiodes (Excelitas) with 30% collection efficiency in the relevant wavelength range, and the detection event was recorded using a time-tagged single photon-counting module (TTM8000).

Data availability statement

The data that support the findings of this study are available from the corresponding authors upon reasonable request.

Author contributions

A. B., R. V. & P. M. A. conceptualized the study. A.B., A. B. P., S. A. I., C. L., H. K., X. Z., T. G., M. S. R. S. & J. L. grew and characterized the films. F. L., M. T. & A. D. did the adhesion and friction force characterizations. T. L. & Y. Z. carried out the refractive index measurement. R. M., C. Y. C. & A. L. G. conducted the single photon emission experiment. A. G. B., M. R. N., D. A. R., P. B. S. & T. I. commented on the manuscript. All the authors discussed the results and contributed to the manuscript preparation.

Conflicts of interest

The authors declare no conflict of interest.

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